

Carbon dot-

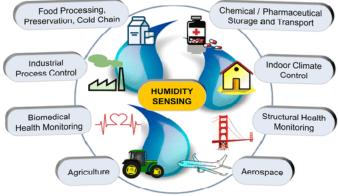
Graphene

2D

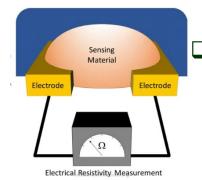
Carbon

Allotropes

Multilayered Graphitic Sheets



**The detection principle of RH sensors** 



Carbon nanostructures (CN)

 Potential advantages of fluoridated carbon

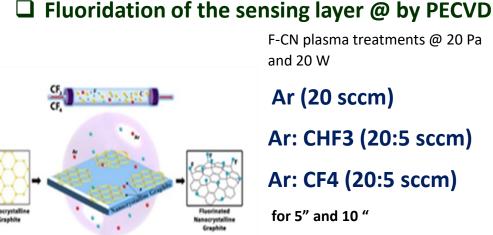
 nanohybrid materials (F-CN) in RH sensors
 High Surface Area

Single-walled Carbon Nanotub

Multi-walled

Carbon Nanotu

- \* Hydrophobicity
- Chemical Stability
- Tunability of Properties
- Compatibility with Flexible Substrates



**Functional Lave** 

**Polymeric Substrate** 

O Adsorp

interdigitated Electrode



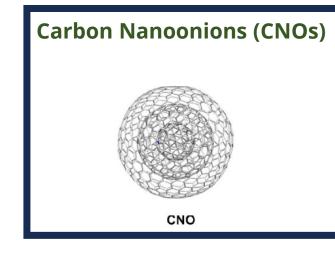
# Exploring Carbon Nanohybrid Materials for Chemoresistive Sensor Sensing Films through Raman Studies



C. Pachiul, O. Simionescul, B. Serbanl, R. Marinescul, N. Dumbravescul, R. Popal, O. Buiul, M. Serbanescu2, Gh. Pristavu2, Gh. Brezeanu2

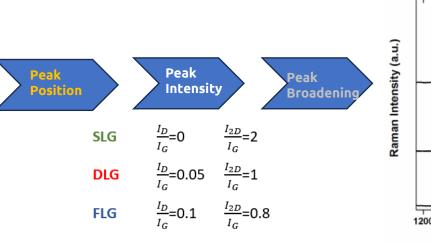
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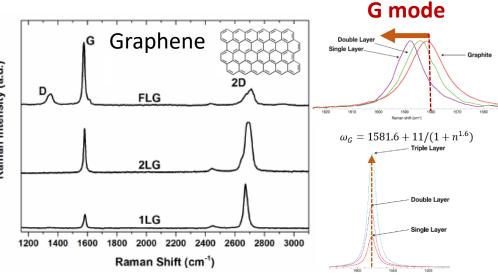
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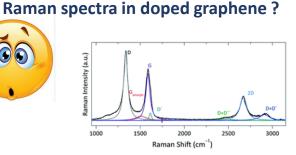


## Advantages of using carbon nanoonions in sensor detection

Large Surface Area **Potential for Multi-Functionalization Electrical Conductivity Gas Adsorption and Sensing** 







- Shift in Raman peak
- Intensity changes

6

- Appearance of new peaks
- D and 2D band evolution



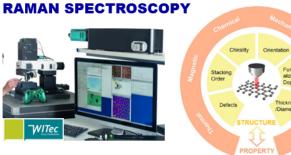
## Exploring Carbon Nanohybrid Materials for Chemoresistive Sensor Sensing Films through Raman Studies ICPAM-15

PAMS-6 November 19 - 26, 2023 Sharm El Sheikh, Egypt

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# **Carbon Nanoonions (CNOs)**



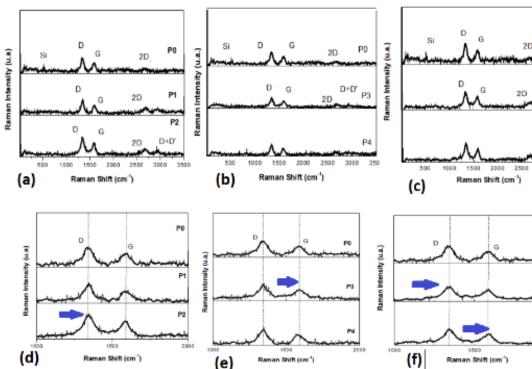
The Raman spectra were recorded with the same system (Witec Alpha 300S/2008 GmbH Germany) using an Nd-YAG laser with 532 nm green excitation

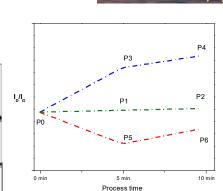
Table 1 Measurement parameters					
spectral range	10 – 3500 cm <sup>-1</sup>				
integration time	20 s				
laser power	1 <u>mW</u>				
grating	600 groves/mm				
laser spot size	400 nm				
spectral resolution	~ 2 cm <sup>-1</sup>				

**Conclusions** 

Samples
P <sub>0</sub> -bulk CNOs
P <sub>1</sub> -F/CNOs : Ar 5"
P <sub>2-</sub> F/CNOs:Ar 10"
$P_{3}F/CNOs:Ar/CHF_{4}5''$
$P_4$ -F/CNOs:Ar/CHF <sub>4</sub> 10"
<b>P</b> <sub>5</sub> _F/CNOs : Ar/CF <sub>3</sub> 5"
P <sub>c</sub> F/CNOs:Ar/CF <sub>2</sub> 10"

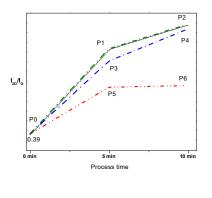
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D+D'

3000



#### □ The position of the D and G bands, shifts slowly to higher values as the fluoridation time increases and the ID/IG ratio increases from 0.39 to NCO to 0.67 for P6 (F-NCO/Ar:CF3) sample.

□ This suggests that the nanocarbonic domains are increasing in width along the hexagonal plane in the material. The width of the lateral crystallite size, La, was determined using the method described in Ref. [1], and the results indicate an increase (from 13.39 nm for NCO to 26.34 nm for P6 (F-NCO/Ar:CF3), as shown in Table).

Sensing films form fluorographene in various degrees of fluorination and the reaction conditions which lead to the hydrophilicity of the sensitive sample of RH humidity sensors.

Sample	<u>WD</u> (cm <sup>-1</sup> )	WG (cm <sup>-1</sup> )	Id/IG	La	I2D/IG
P <sub>0</sub> -bulk CNOs	61.8	43.06	1.435	13.39	0.390
P <sub>1</sub> -F-CNOs:Ar 5"	71.88	49.8	1.443	13.31	0.607
P2.F-CNOs:Ar 10"	90.38	62.32	1.450	13.25	0.669
P <sub>3</sub> .F-CNOs:Ar/CHF <sub>4</sub> 5"	64.95	40.19	1.616	11.89	0.575
P <sub>4</sub> .F-CNOs:Ar/CHF <sub>4</sub> 10"	65.94	39.53	1.668	32.06	0.657
P <sub>5</sub> .F-CNOs:Ar/CF <sub>3</sub> 5 <sup>°°</sup>	83.11	63.63	1.306	14.71	0.509
P <sub>6</sub> .F-CNOs:Ar/CF <sub>3</sub> 10"	90.38	65.95	1.370	26.34	0.514

Thank you for your attention.



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